

NAME: _____ Score _____/10

Please **print** your name**SHOW ALL YOUR WORK IN A NEAT AND ORGANIZED FASHION**

On this quiz I will lead you through a proper way of writing a solution to a question. You are to supply the details by filling in each blank. **No computations are required or even desired.**

Problem: What quantity of a 60% acid solution must be mixed with a 30% acid solution to produce 300 mL of a 50% acid solution ?

(*) and (**) should not be the same.

Analysis:

Let x be the amount (measured in milliliters) of **60% solution** to be added.

The volume of the final mixture will be **300** mL.

The **amount of acid in the final solution** is **$(0.5)(300)$** . (*)

The amount of acid contributed by the 60% solution is **$(0.6)x$** .

The amount of the 30% solution will be **$300 - x$** mL.

The amount of **acid** contributed by the 30% solution is $(0.3)(300 - x)$.

The **amount of acid in the final solution** is **$(0.6)x + (0.3)(300 - x)$** . (**)

We now have the amount of **acid** in the **final** solution written in two ways.

Therefore the mathematical model for this concentration problem is the linear equation in one variable

(Insert the Model/equation here) **$(0.6)x + (0.3)(300 - x) = (0.5)(300)$**

Solution: **(I have solved the equation so you don't need to)**

Ordinary methods now may be used to solve this equation to obtain $x = 200$.

Conclusion: **(Must be based on the correct solution as stated above.)**

200 milliliters of **60% solution** must be added to 100 milliliters of **30% solution** to obtain

300 milliliters of **50% solution**.